

Lessons learned from long-term corrosion investigations and –monitoring in saline thermal waters

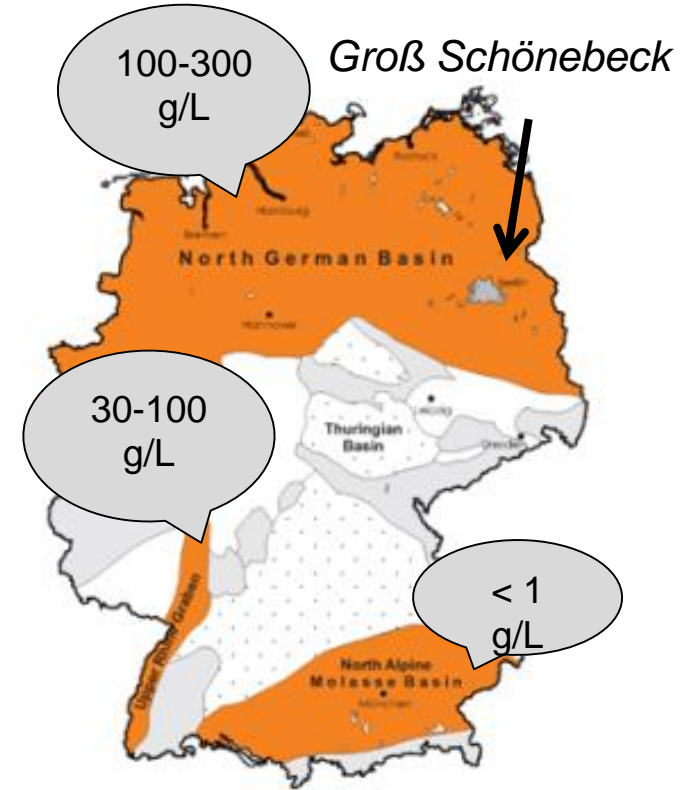
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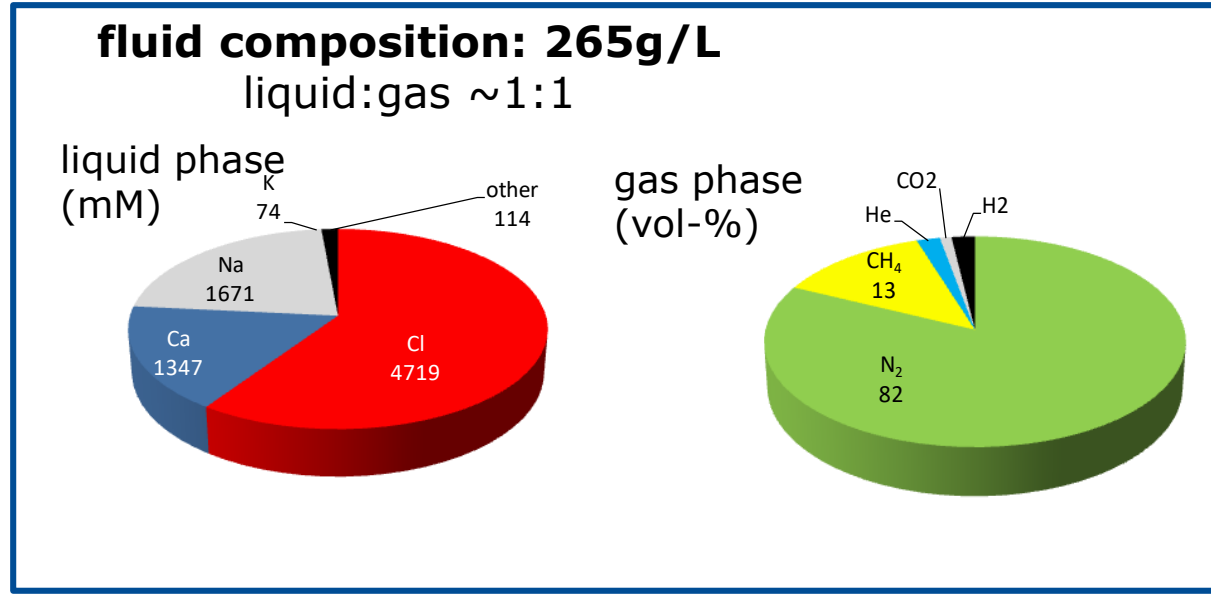
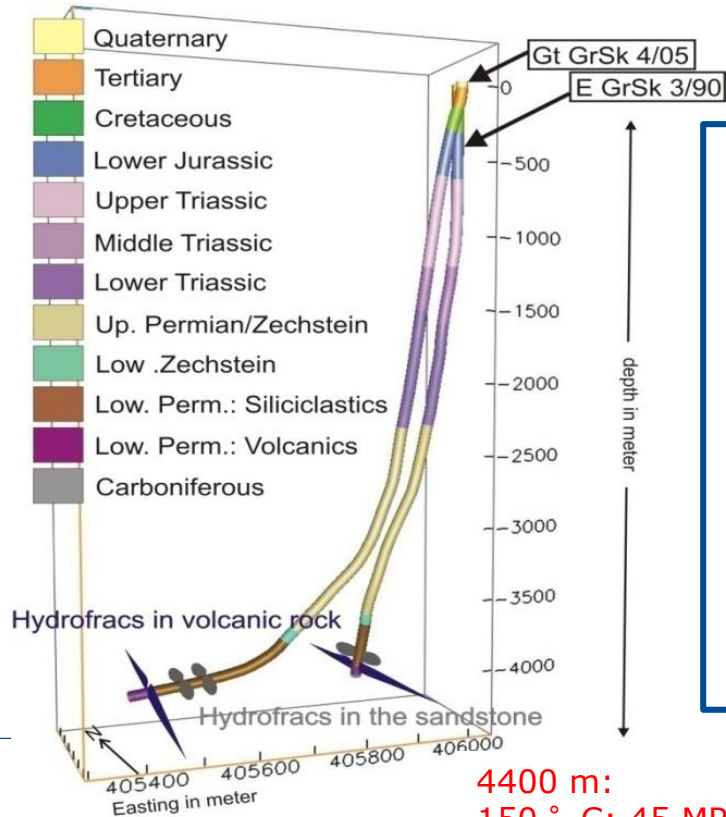
Corrosion in saline waters

Geothermal fluids are corrosive.
High salinity increases corrosion of
steel components.

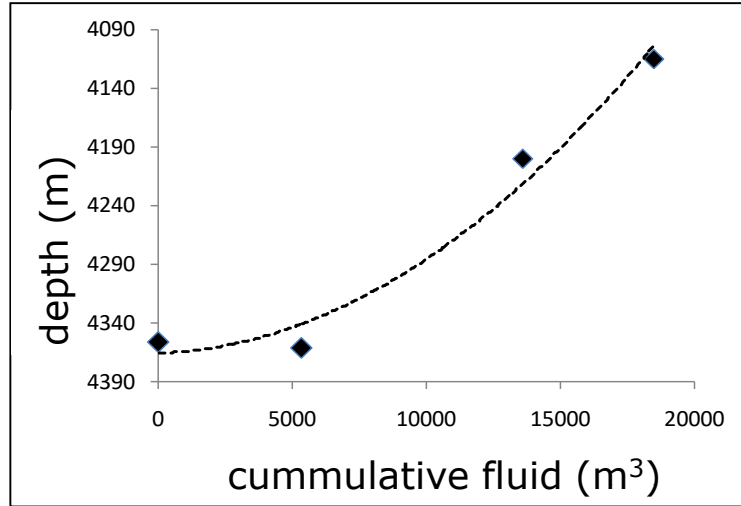
Considerations on material selection
before setting up a geothermal well and
installations



Background: The geothermal research platform Groß Schönebeck



Groß Schönebeck circulation tests (2011-2013)



1. Reduced production rate and change of total depth over time.

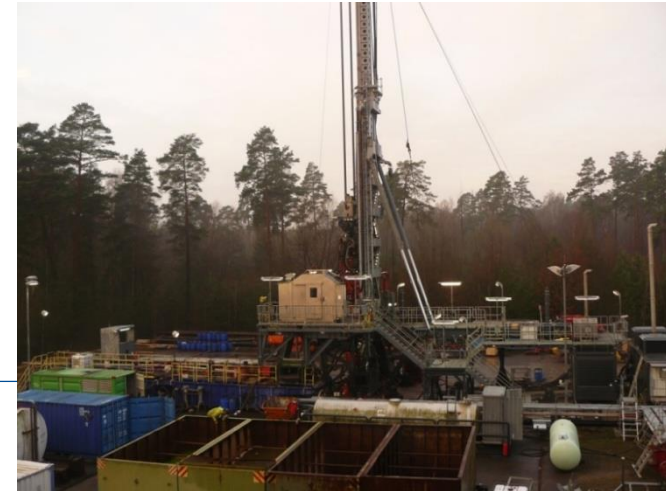


2. Bailer sampling

indicated: filling of the production well (native copper, magnetite, barite...)

3. Removal of the fill:

1. Coiled tubing
2. Work-over rig



Chemical monitoring – during and after circulation tests

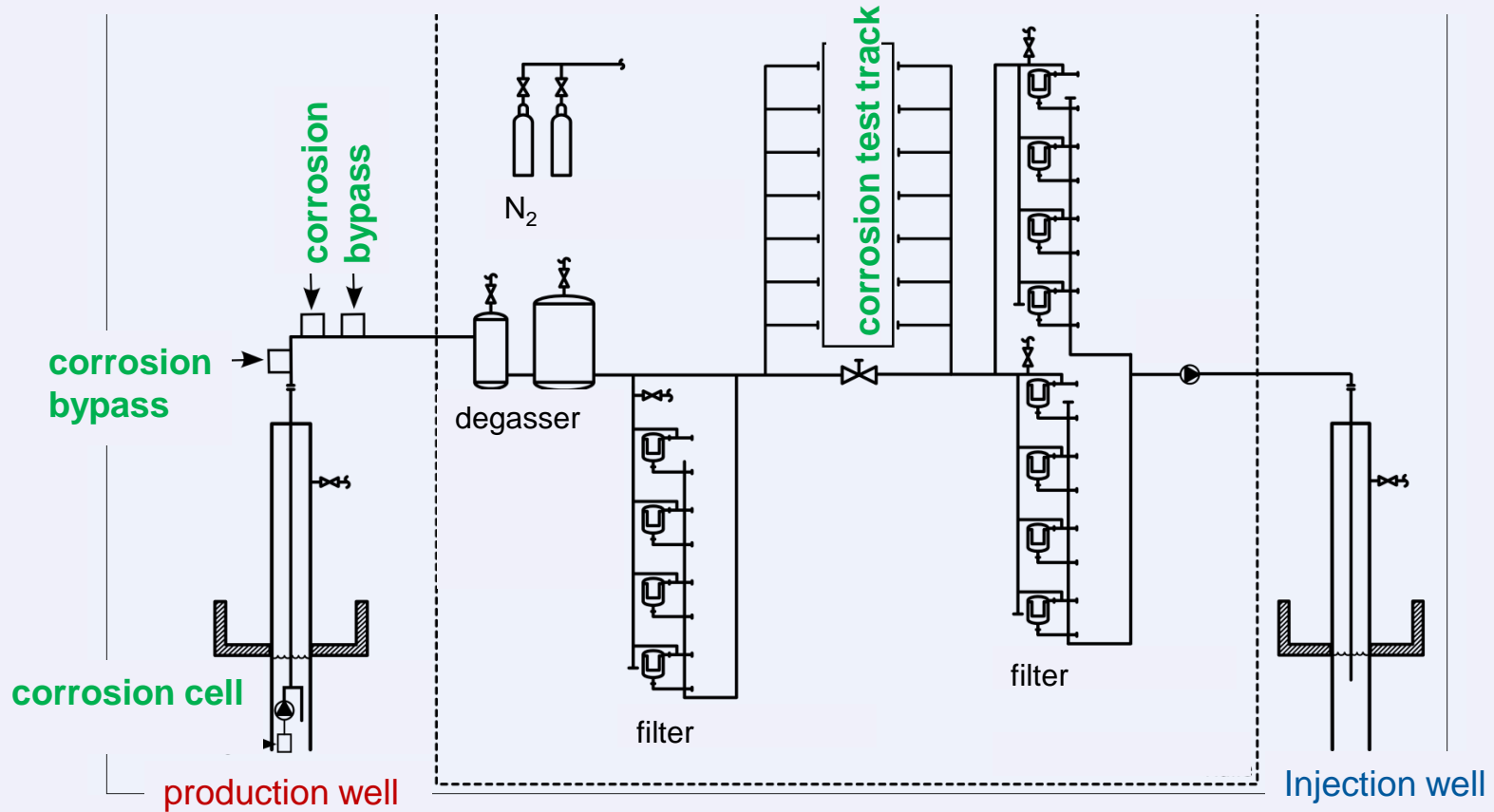
- Thermal water composition
Feldbusch et al., 2013; Regensburg et al. 2016
- Gas composition
Feldbusch et al., 2018
- Solid phase composition
Regensburg et al., 2015
- **Corrosion**



Material protection: epoxy resin (Vetco TK236)

- Non-polar, electric isolating coating
- Temperature resistivity: 200 ° C
- thickness: < 250 µm

Corrosion monitoring



Corrosion monitoring

1. Corrosion test track (bypass)
2. Component bypass
3. Corrosion cell (below production pump)

} planned

4. Coating instability (tubing)
5. Wellbore: electrochemical corrosion

} unexpected

1. Monitoring corrosion test track

S+C materials (cast and modeling treatment)

alloy	corrosion
G 45 Mo	no pitting corrosion
G 45 Mo mod.	
A59	
A 254 SMO	matted but no pitting
A625	
A825	
A31	
G625	
316 L	
C80L	general corrosion

corrosion resistant

limited stable

unstable



2. Components (bypass)



2. Results components (bypass)

alloy	corrosion
Duplex cast material 1.4469	No corrosion
Superduplex 1.4501	
Austenitic cast steel 1.4408	
CrNiMo steel 1.4418	
Carbon steel parts	general corrosion (< 0.2 mm/a)

corrosion resistant

Little corrosion



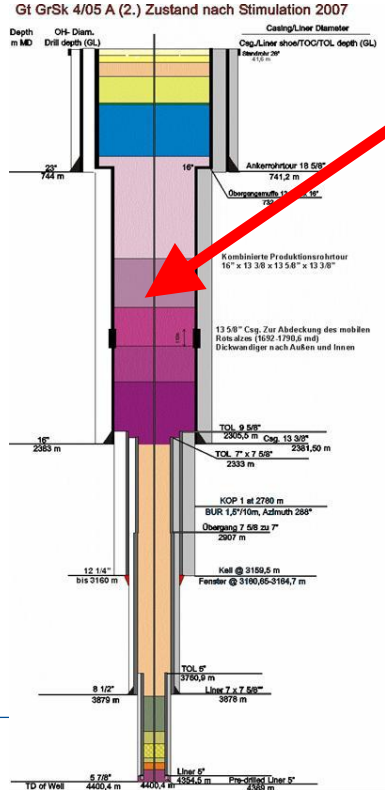
vane housing

3. Corrosion test cell below the production pump

Production well

ESP in 1200 m

Corrosion test cell



Test cell

3. Results: corrosion-tests below the pump

- austenitic CrNiMo-steel A
- austenitic CrNiMo- Stahl B
- titanium gr. 12
- alloy 625
- alloy 59
- alloy 31
- alloy G45 Mo (7%)
- alloy G45 Mo (9,5%)



Example sample after 3 years exposure

→ all samples were corrosion-resistant
→ possibly corrosion protection by scaling

Unexpected corrosion reactions

4. Coating instability: Epoxy resin (Vetco TK236)

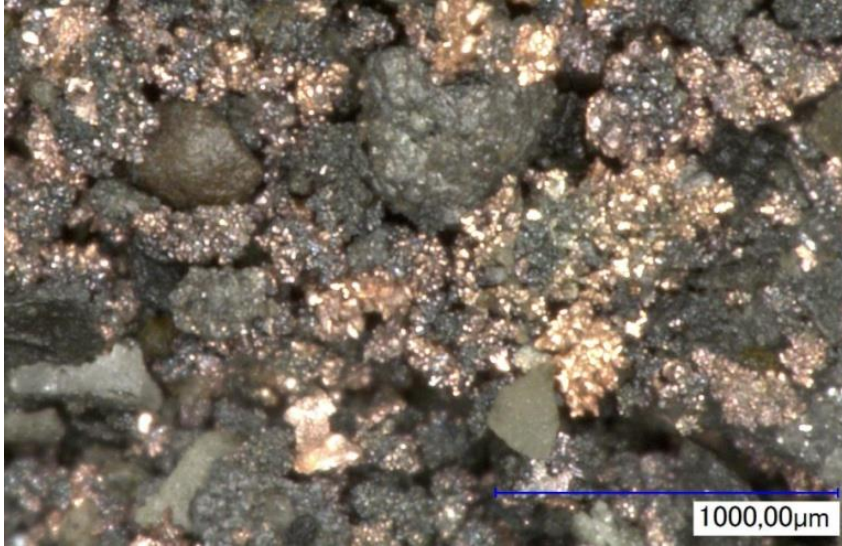


Tubing above ground: scaling (*barite*), intact coating; adherence > 20,68 Mpa

Tubing (<1200m) below ground:
some tubes: coating removal

→ decreasing adhesion between topcoat and primer (red) → defect of manufacturing

5. Electrochemical corrosion in the well bore/ liner wall



Corrosion products: High amounts of

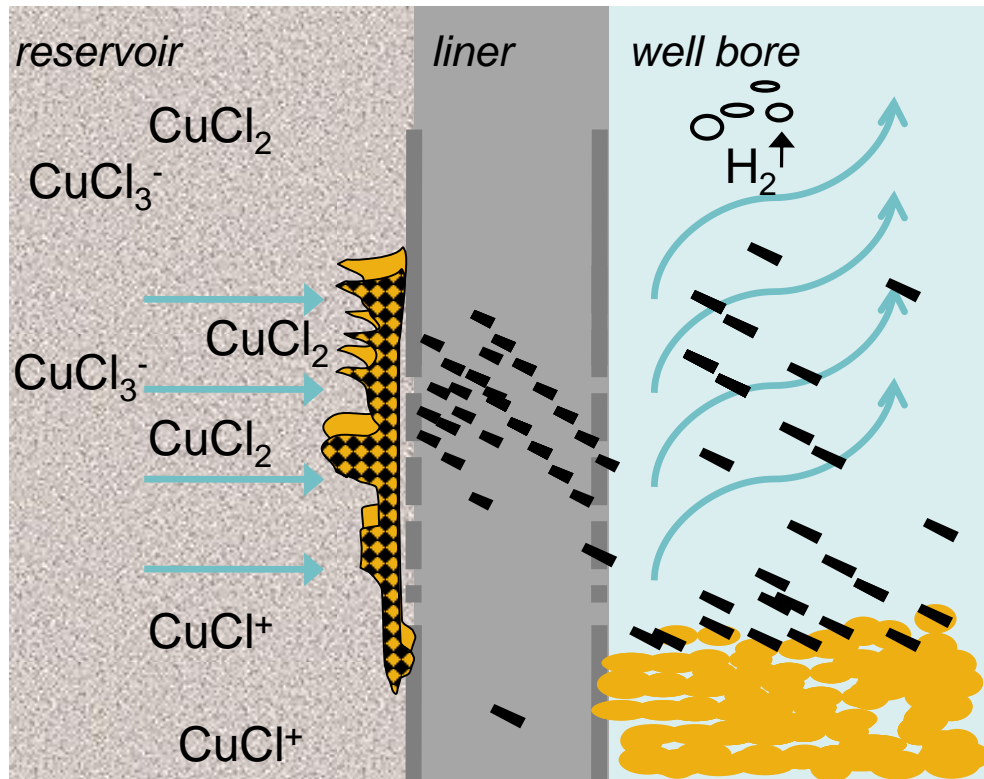
- native copper (Cu)
- magnetite (Fe_3O_4)

found at well bottom

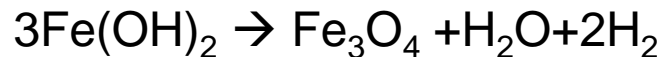
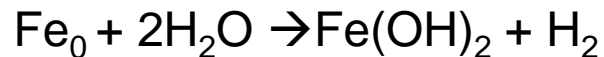
- hydrogen gas (H_2)

measured in the gas phase

5. Electrochemical corrosion in the well bore



Corrosion process:



Summary monitoring

- Most tested materials (apart of carbon steel) proved to be corrosion resistant within the installations.
- Stability of coatings cannot be guaranteed.
- Main risk: Electrochemical corrosion of carbon steel casing with dissolved Cu.

Corrosion prevention strategy

Use of higher alloyed casing:

- corrosion resistant (T, salinity, Cu²⁺)
- available for casing
- economically feasible

Cladded tubes

- Combining properties of C-steel (stability and ductility) with corrosion resistivity of highly alloyed steels
- Cladding of C-steel inside and/or outside by mechanical or metallurgical technique

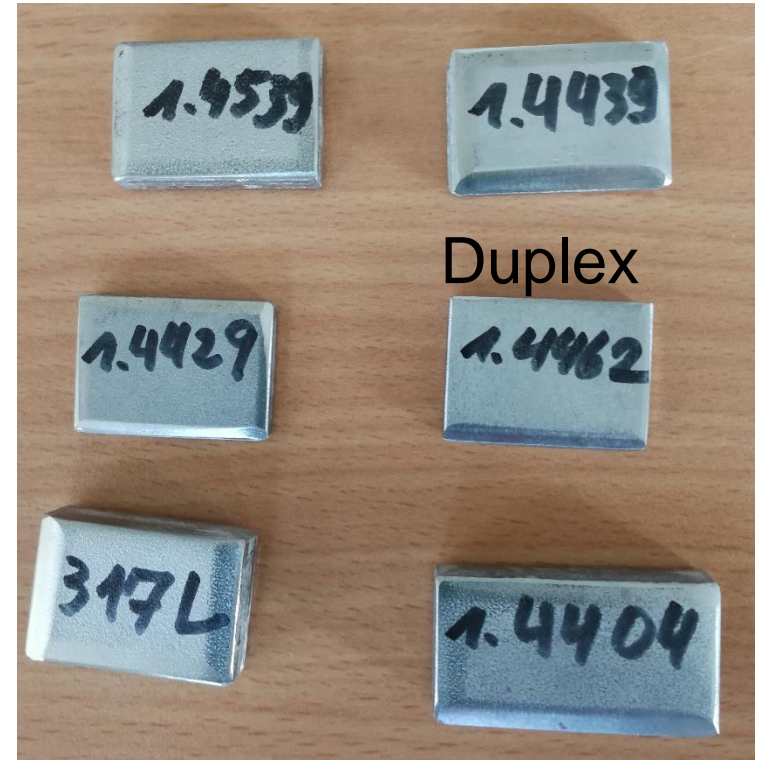
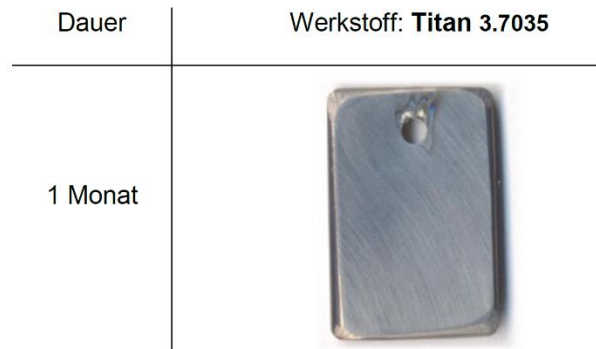


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Tests of cladding materials in autoclaves:

150° C, 1 mM Cu, 1.5 M CaCl₂, 2 M NaCl

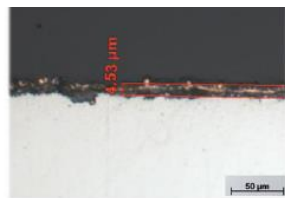
- 5 Austenitic steel
 - 1 Duplex
 - Titanium
 - C-steel (St37)
- } Used for cladding



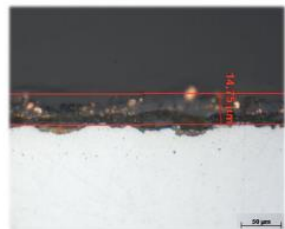
Results

C-Stahl

24 h



7 days



Austenites and Duplex

Dauer

1.4438

1.4404

1.4429

1.4439

1.4462

1.4539

7
e
days



1
month



→ Reaction of Cu^{2+} with C-Steel increases over time

→ Highly alloyed steels are corrosion resistant (slight crevice corrosion)

Outlook

Groß Schönebeck: → Need of pipe and casing materials resistant against electrochemical corrosion with Cu^+ , Cu^{2+}

e.g. cladded pipes

Thank you



Gefördert durch:



Bundesministerium
für Wirtschaft
und Energie



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Bayer Technology Services



aufgrund eines Beschlusses
des Deutschen Bundestages



References

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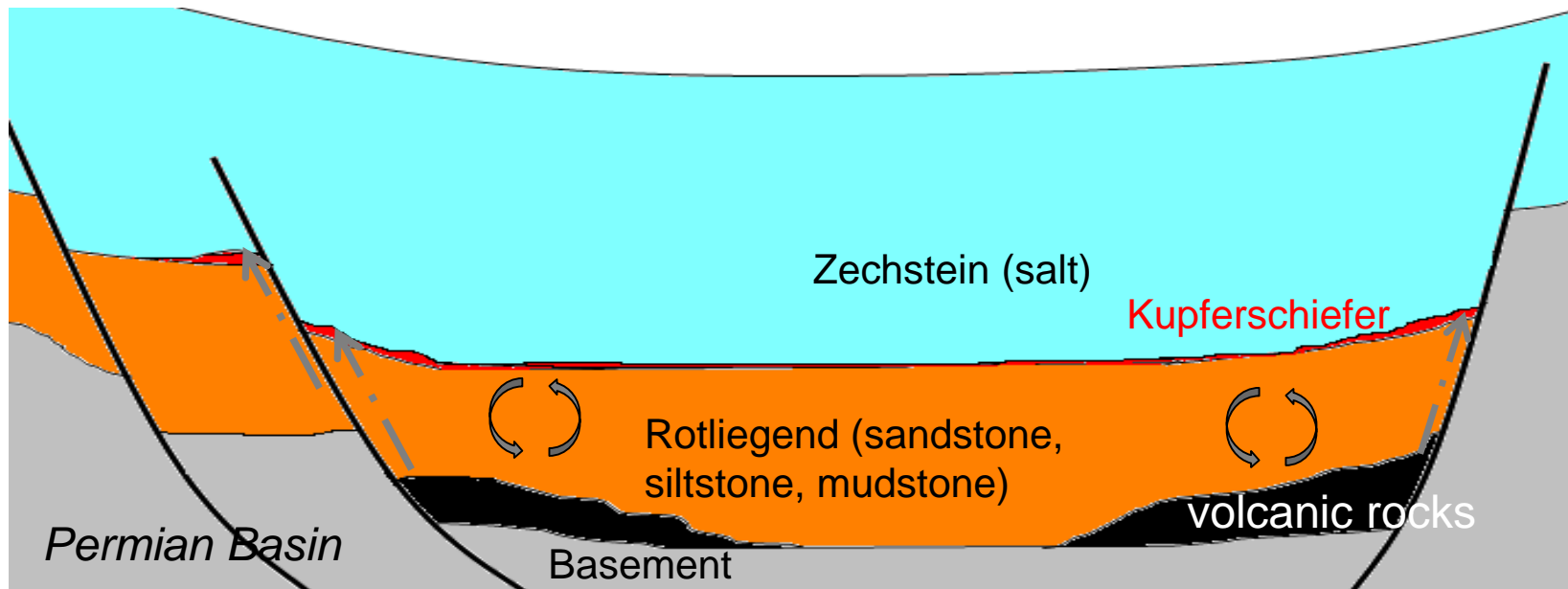
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Woher stammt das Kupfer im Rotliegend?

1. Vulkangesteine als Quelle der Schwermetalle(Cu, Pb, Hg, Zn)
2. Auslaugung der Schwermetalle durch saline Rotliegendfluide entlang Störungszonen (Bildung von Chloridekomplexen)



Total removed amount of solids (filter, bailer, well cleaning): ~ 600 L

Mean composition of all samples:

